

## Multiple Oral Dose Administration

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### Objectives

- To recognize and use the integrated equations for multiple oral dose administration
- To calculate appropriate multiple dose regimen
- To define, calculate and use the parameter  $\bar{C}_p$
- To use the superposition principle to calculate  $C_p$  after non uniform dosing regimen

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### Oral Dose Administration

#### - Single Dose

$$C_p = \frac{F \cdot Dose \cdot ka}{V \cdot (ka - kel)} \cdot [e^{-kel t} - e^{-ka t}]$$

#### - Multiple Dose

$$C_p = \frac{F \cdot Dose \cdot ka}{V \cdot (ka - kel)} \cdot \frac{1 - e^{-n \cdot kel t}}{1 - e^{-kel t}} \cdot e^{-kel t} - \frac{1 - e^{-n \cdot ka t}}{1 - e^{-ka t}} \cdot e^{-ka t}$$

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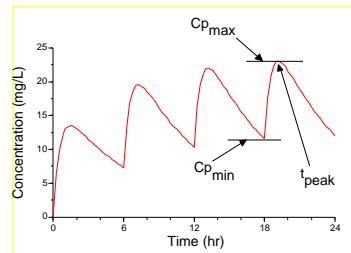


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### C<sub>p</sub> versus Time



### C<sub>p</sub><sub>max</sub> and C<sub>p</sub><sub>min</sub>

- C<sub>p</sub><sub>max</sub> could be calculated if t<sub>max</sub> is known from 'full' equation
- C<sub>p</sub><sub>min</sub> can be calculated at t = 0 or t =

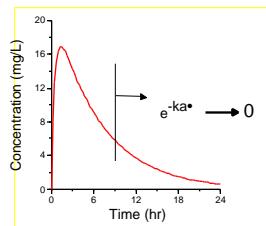
### C<sub>p</sub><sub>min</sub> Equation

at t = 0 and with n large ->  
 $e^{-n \cdot k_{el} \cdot t}$  and  $e^{-n \cdot k_a \cdot t}$  both -> 0

$$C_{p\min} = \frac{F \cdot \text{Dose} \cdot k_a}{V \cdot (k_a - k_{el})} \cdot \frac{1}{1 - e^{-k_{el} \cdot t}} - \frac{1}{1 - e^{-k_a \cdot t}}$$

## Simplify Equation

if  $e^{-ka^*} \gg 0$




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## Simplify Equation...

$$C_{p_{\min}} = \frac{F \cdot \text{Dose} \cdot ka}{V \cdot (ka - kel)} \cdot \frac{e^{-kel^*}}{1 - e^{-kel^*}}$$

or

$$C_{p_{\min}} = A \cdot \frac{R}{1 - R}$$

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## Accumulation

$$\frac{C_{p_{\min}}}{C_{p_1}} = \frac{\frac{F \cdot \text{Dose} \cdot ka}{V \cdot (ka - kel)} \cdot \frac{e^{-kel^*}}{1 - e^{-kel^*}}}{\frac{F \cdot \text{Dose} \cdot ka}{V \cdot (ka - kel)} \cdot [e^{-kel^*}]}$$

or

$$\frac{C_{p_{\min}}}{C_{p_1}} = \frac{A \cdot \frac{R}{1-R}}{A \cdot R} = \frac{1}{(1-R)} = \frac{1}{1 - e^{-kel^*}}$$

**if  $e^{-ka^*} \approx 0$**

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Loading Dose =

$$\text{Loading Dose} = \frac{\text{Maintenance Dose}}{(1 - R)}$$

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### Further Simplification

- if  $ka \gg kel$

$$\frac{ka}{(ka - kel)} \approx 1$$

$$C_{p\min} = \frac{F \cdot \text{Dose}}{V} \cdot \frac{e^{-kel \cdot t}}{1 - e^{-kel \cdot t}}$$

- this is an even more extreme approximation
- if  $ka$  unknown but absorption fast it may be useful

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### The Three Equations

F	1	
Dose	250	mg
ka	2	hr <sup>-1</sup>
kel	0.15	hr <sup>-1</sup>
V	15	L
Tau	6	hr
	C <sub>p</sub> min	
Full Equation	12.3444	
ka • tau >>	12.3445	
ka >> kel	11.4186	

EXCEL

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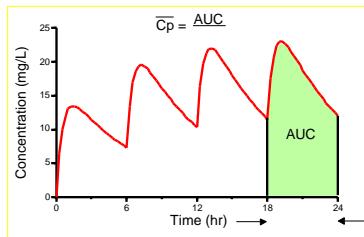
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## Average Concentration, $\bar{C}_p$

- Average plasma concentration during steady state can be useful
- Calculated as AUC/

$$\bar{C}_p = \frac{\int C_p \cdot dt}{V \cdot k_{el}}$$

## Graphical Representation



## Average Concentration

$$\bar{C}_p = \frac{\int C_p \cdot dt}{V \cdot k_{el}} = \frac{F \cdot \text{Dose}}{V \cdot k_{el}}$$

- Note Changing the Dose and the Dosing Interval by the Same Factor gives the SAME  $\bar{C}_p$
- e.g. 300 mg q12h gives the same  $\bar{C}_p$  as 100 mg q4h
  - however  $C_{p_{min}}$  and  $C_{p_{max}}$  would change

## Example Calculation

- Data: F = 1; V = 30 L;  $t_{1/2} = 6 \text{ hr}$ ; kel = 0.116  $\text{hr}^{-1}$
- Question: Dose every 12 hours to achieve 15 mg/L
- Equation:  $\bar{C}_p = \frac{F \cdot \text{Dose}}{V \cdot \text{kel}}$   
 $\text{Dose} = \frac{\bar{C}_p \cdot V \cdot \text{kel}}{F}$   
 $\text{Dose} = \frac{15 \times 30 \times 0.116 \times 12}{1} = 624 \text{ mg}$

## Loading Dose

$$R = e^{-\text{kel} \cdot t} = e^{-0.116 \times 12} = 0.25$$

$$\text{Loading Dose} = \frac{\text{Maintenance Dose}}{1 - R}$$

$$\text{Loading Dose} = \frac{624}{1 - 0.25} = 832 \text{ mg}$$

- Dosing Regimen 800 - 850 mg to start then 600 - 650 mg every 12 hours

## Estimate $C_p_{\min}$ and $C_p_{\max}$

- Assuming  $k_a \gg \text{kel}$  and  $e^{-k_a t} \rightarrow 0$

$$C_p_{\min} = \frac{1 \times 624}{30} \times \frac{0.25}{1 - 0.25} = 6.93 \text{ mg/L}$$

$C_p_{\max} \approx 15 + (15 - 7) = 23 \text{ mg/L}$   
 $C_p = 15 \text{ mg/L}$   
 $C_p_{\min} = 7 \text{ mg/L}$

## Alternative Regimen

312 mg q 6 h

$$C_{P_{\min}} = \frac{1 \times 312}{30} \times \frac{0.5}{1 - 0.5} = 10.4 \text{ mg/L}$$

$$- \text{Cp}_{\max} \approx 15 + (15 - 10) = 20 \text{ mg/L}$$

$$\overline{C_p} = 15 \text{ mg/L}$$

$$C_{P_{\min}} = 10 \text{ mg/L}$$

## Non-Uniform Dosing

## Superposition Principle

- Applies when all disposition processes are linear
    - disposition = DME
    - linear = first order
  - Allows addition of concentrations from separate doses = multiple dose
  - Double the dose = double the concentration

## Equations

$$Cp_1 = \frac{200}{15} \times e^{-0.15 \times t}$$

$$Cp_2 = \frac{300}{15} \times e^{-0.15 \times (t-6)}$$

$$Cp_3 = \frac{100}{15} \times e^{-0.15 \times (t-18)}$$

$$C_p = C_{p_1} + C_{p_2} + C_{p_3} \quad AL (24 \text{ hr})$$

## Equations

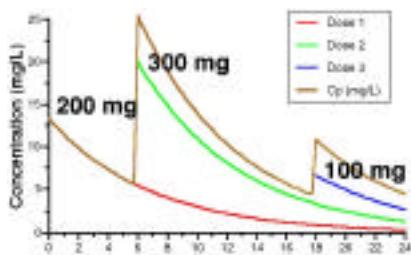
$$Cp_1 = \frac{200}{15} \times e^{-0.15 \times t} = 0.36 \text{ mg/L}$$

$$Cp_2 = \frac{300}{15} \times e^{-0.15 \times (t-6)} = 1.34 \text{ mg/L}$$

$$Cp_3 = \frac{100}{15} \times e^{-0.15 \times (t-18)} = 2.71 \text{ mg/L}$$

$$Cp = Cp_1 + Cp_2 + Cp_3 = 4.41 \text{ mg/L}$$

## Graphical Representation



## Another Approach

$$Cp_1^0 = \frac{200}{15} = 13.33 \text{ mg/L}$$

$$Cp_1^6 = 13.33 \times e^{-0.15 \times 6} = 5.42 \text{ mg/L}$$

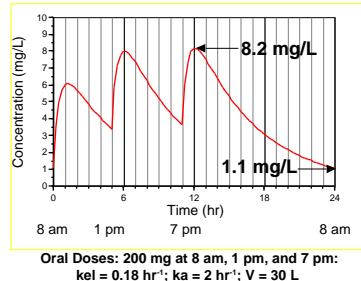
$$Cp_2^0 = 5.42 + \frac{300}{15} = 25.42 \text{ mg/L}$$

$$Cp_2^{12} = 25.42 \times e^{-0.15 \times 12} = 4.20 \text{ mg/L}$$

$$Cp_3^0 = 4.20 + \frac{100}{15} = 10.87 \text{ mg/L}$$

$$Cp_3^6 = 10.87 \times e^{-0.15 \times 6} = 4.42 \text{ mg/L}$$

### Multiple Oral Doses at steady state



### Non-Uniform Dosing

May be acceptable if

- Drug has wide therapeutic index
- No therapeutic disadvantage to low overnight plasma concentration
  - e.g. analgesic if patient asleep

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