

Nonlinear Pharmacokinetic Models

- Objectives:
- To understand the schemes and differential equations associated with nonlinear pharmacokinetic models
 - To understand the effect of parallel pathways
 - To estimate the parameters K_m and V_m
 - To design appropriate dosage regimen for drugs with nonlinear elimination

- Nonlinear Processes
- Lower concentration > first order
 - Higher concentration > zero order
 - Concentration or dose dependent kinetics
 - Enzyme reaction associated with metabolism may be saturable
 - Enzyme reaction may have a maximum rate limited by substrate
 - Basic enzyme kinetics have application to pharmacokinetics

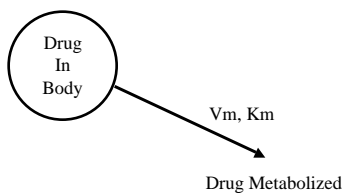
Michaelis-Menten Kinetics

$$\text{Rate of Elimination} = \frac{V_m \cdot C_p}{K_m + C_p}$$

- where V_m is the maximum rate of metabolism
- and K_m is Michaelis constant, the concentration (or amount) of drug at which the rate is 1/2 maximum

Scheme

- MM only elimination pathway



Differential Equation

- Single elimination pathway

$$\frac{dC_p}{dt} = -\frac{V_m \cdot C_p}{K_m + C_p}$$

Equation at Low Concentrations

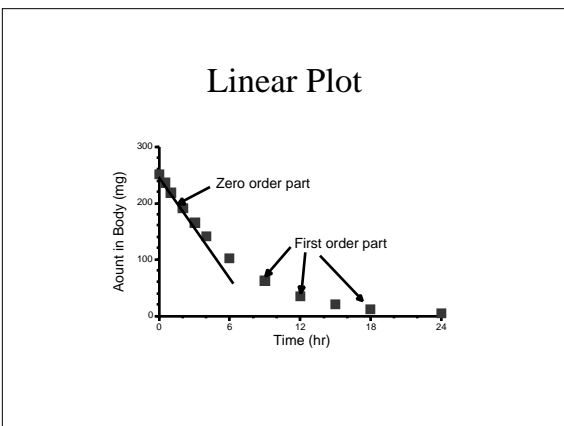
- $K_m > C_p$
- $K_m + C_p \approx K_m$
- Therefore

$$\frac{dC_p}{dt} = -\frac{V_m \cdot C_p}{K_m} = -k' \cdot C_p$$
- pseudo first order elimination

Equation at High Concentration

- $C_p > K_m$
- $K_m + C_p \approx C_p$
- Therefore

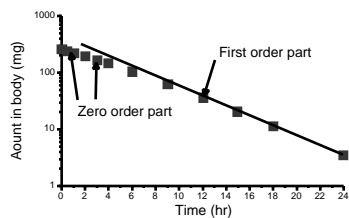
$$\frac{dC_p}{dt} = -\frac{V_m \cdot C_p}{C_p} = -V_m$$
- zero order elimination



High Dose - Concentration

- Slope constant on linear graph == zero order
- Slope approaches $-V_m$

Semi-log Plot



Low Dose - Concentration

- Slope constant on semi-log graph == first order
- Slope approaches $-V_m/K_m$

Example - Phenytoin

- Average K_m 4 mg/L (1 - 15 mg/L)
- Average V_m = 500 mg/day (100 - 1000 mg/day)
- Therapeutic window 10 - 20 mg/L (total C_p)
- Overdose possible if dose adjustment is not appropriate
- Half-life at low doses 12 hr, maybe greater than 24 hr at higher doses
- From 25 to 23 mg/L in 24 hours (cf. $25 > 12.5 > 6$ mg/L when $t_{1/2}$ is 12 hr)

Evans, Schentag, Jusko. *Applied Pharmacokinetics*, Applied Therapeutics, Vancouver, WA 1992

Effect of MM Kinetics on $t_{1/2}$

- $t_{1/2}$ larger as concentration increases; i.e. slower elimination

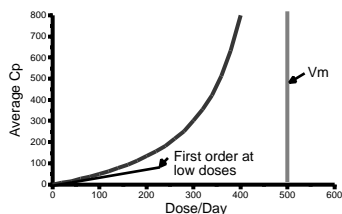
$$\frac{dC}{dt} = -k_e I \cdot C = -\frac{V_m \cdot C}{K_m + C}$$

$$\text{since } k_{el} = \frac{0.693}{t_{1/2}}$$

$$\frac{0.693}{t_{1/2}} = \frac{V_m}{K_m + C}$$

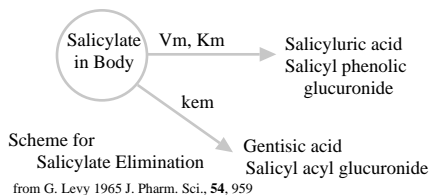
$$t_{1/2} = \frac{0.693 \cdot (K_m + C)}{V_m}$$

Effect of MM Kinetics on \bar{C}_p

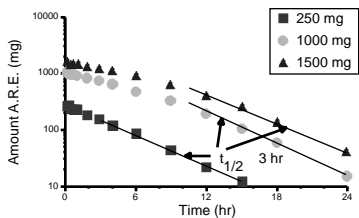


Parallel Pathways

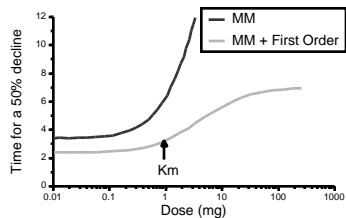
- Linear and Nonlinear Elimination



Salicylate Elimination



Parallel Pathways



Adapted from Niazi, S. 1979 "Textbook of Biopharmaceutics and Clinical Pharmacokinetics, Appleton-Century Croft, New York, NY Fig 7.14, page 181

Parallel Pathways

- At low dose k_{em} and V_m/K_m is larger, thus $t_{50\%}$ is smaller
- At higher doses effective (pseudo first order) rate constant for MM process is small, thus $t_{50\%}$ is larger

Dosing Approaches

First Dose

- Use population (average) values for phenytoin
- $V_m = 7 \text{ mg/kg/day}$ and $K_m = 5 \text{ mg/L}$
- Aim for $C_p = 15 \text{ mg/L}$ in 80 kg patient
- Equation:
$$\text{Dose Rate} = \frac{V_m \cdot \bar{C}_p}{K_m + \bar{C}_p}$$

$$= \frac{7 \times 80 \times 15}{(5 + 15)} = 420 \text{ mg/day}$$

PDR Recommends 300 mg/day - probably better to start low

Determine Second Dose Regimen

Give an initial dosage regimen and measure C_p

- For example if after 420 mg/day, \bar{C}_p is 20 mg/L
- Assume $K_m = 5 \text{ mg/L}$ and calculate V_m

$$V_m = DR + \frac{DR \cdot K_m}{\bar{C}_p} = 420 + \frac{420 \times 5}{20} = 525 \text{ mg/day}$$

$$\text{Dose Rate} = \frac{V_m \cdot \bar{C}_p}{K_m + \bar{C}_p} = \frac{525 \times 15}{5 + 15} = 394 \text{ mg/day}$$

Change in Dose

- Note 420 mg/day >> 20 mg/L
- and 394 mg/day >> 15 mg/L
- 6% increase in dose results in 25% increase in concentration

A Nomogram

From Winter, M.E. and Tocco, T.N. (1986) Cl₁₆ Phenytoin in Applied Pharmacokinetics, Ed Evans, W.E., Schornag, J.J., and Jusko, W.J., 2nd, Applied Therapeutics, Fig 16-11, page 513

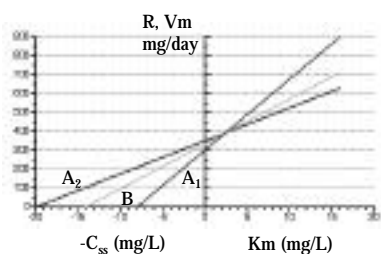
Using a Nomogram

- Line A from C_{pss} = 8 mg/L at 4.3 mg/kg/day (= 300 mg/day for 70 kg)
- Line B from midpoint of shape to new C_{pss} = 15 mg/L
- New Dose Rate is 5.2 mg/kg/day (= 364 mg/day)

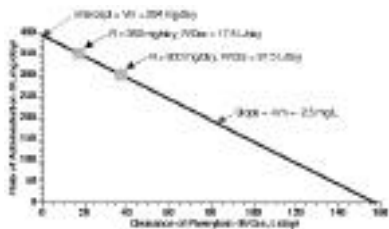
After Two or More Concentrations

- Determined at Steady State (note extended half-life)
- For Both (All) Concentrations
 - Good patient compliance
 - Steady State is Achieved
 - Protein Binding is unchanged
- Example
 - 300 mg/day -> 8 mg/L
 - 350 mg/day -> 20 mg/L

Graphical Method - 1



Graphical Method - 2



A Third Method

- After two previous dosage regimen
- For example, $C_{p1} = 8 \text{ mg/L}$, $C_{p2} = 20 \text{ mg/L}$,
 $R_1 = 300 \text{ mg/day}$, and $R_2 = 350 \text{ mg/day}$

$$DR_1 = \frac{V_m \cdot \overline{C_{p1}}}{K_m + C_{p1}}$$

$$300 = \frac{V_m \cdot 8}{K_m + 8} \quad \text{and} \quad 350 = \frac{V_m \cdot 20}{K_m + 20}$$

Solve Simultaneous Equations

$$300 \cdot K_m + 300 \times 8 = V_m \cdot 8$$

$$350 \cdot K_m + 350 \times 20 = V_m \cdot 20$$

$$300 \times 350 \cdot K_m + 300 \times 350 \times 8 = 350 \times 8 \cdot V_m$$

$$300 \times 350 \cdot K_m + 300 \times 350 \times 20 = 300 \times 20 \cdot V_m$$

$$300 \times 350 \times (20 - 8) = (300 \times 20 - 350 \times 8) \cdot V_m$$

$$V_m = \frac{1260000}{3200} = 394 \text{ mg/day}$$

$$K_m = \frac{8 \cdot V_m - 300 \times 8}{300} = \frac{750}{300} = 2.5 \text{ mg/L}$$

Graphical Methods

- With more than two previous dosage regimens the graphical methods can be used with more data points or lines plotted.

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